## Quantum Mechanics of Atoms and Molecules (PC3602) Exercise 2

- 1. A particle has spin quantum number s = 1.
  - (a) What are the eigenvalues of  $\hat{S}_z$ ?
  - (b) Using the eigenstates of  $\hat{S}_z$  as basis, determine in matrix form its operators  $\hat{S}_z$ , and  $\hat{S}^{\pm}$ .
  - (c) Determine the eigenvalues and eigenstates of  $\hat{S}_y$ .
- 2. (a) Prove any one of the following three commutation relationships

$$\begin{bmatrix} \hat{L}_x, \ \hat{L}_y \end{bmatrix} = i\hbar\hat{L}_z, \quad \begin{bmatrix} \hat{L}_y, \ \hat{L}_z \end{bmatrix} = i\hbar\hat{L}_x, \quad \begin{bmatrix} \hat{L}_z, \ \hat{L}_x \end{bmatrix} = i\hbar\hat{L}_y.$$

(b) Use (a) to prove

$$\begin{bmatrix} \hat{L}_z, \ \hat{L}^{\pm} \end{bmatrix} = \pm \hbar \hat{L}^{\pm}, \quad \begin{bmatrix} \hat{L}^+, \ \hat{L}^- \end{bmatrix} = 2\hbar \hat{L}_z$$

where  $\hat{L}^{\pm} = \hat{L}_x \pm i\hat{L}_y$  are defined as raising and lowering operators for angular momentum.

(c) Prove

$$\hat{L}^{+}\hat{L}^{-} = \hat{L}^{2} - \hat{L}_{z}^{2} + \hbar\hat{L}_{z}, \quad \hat{L}^{-}\hat{L}^{+} = \hat{L}^{2} - \hat{L}_{z}^{2} - \hbar\hat{L}_{z}$$

where  $\hat{L}^2 = \hat{\mathbf{L}}^2 = \hat{L}_x^2 + \hat{L}_y^2 + \hat{L}_z^2$ .

3. (a) Prove the following commutation relationships

$$\begin{bmatrix} \hat{L}^2, \ \hat{L}_1^2 \end{bmatrix} = \begin{bmatrix} \hat{L}^2, \ \hat{L}_2^2 \end{bmatrix} = \begin{bmatrix} \hat{L}^2, \ \hat{L}_z \end{bmatrix} = 0$$
$$\begin{bmatrix} \hat{L}_z, \ \hat{L}_1^2 \end{bmatrix} = \begin{bmatrix} \hat{L}_z, \ \hat{L}_2^2 \end{bmatrix} = 0$$

where  $\hat{\mathbf{L}} = \hat{\mathbf{L}}_1 + \hat{\mathbf{L}}_2$  is the total angular momentum,  $\hat{L}_z = \hat{L}_{z1} + \hat{L}_{z2}$  is its zcomponent, and we have assumed  $[\hat{L}_1^2, \hat{L}_2^2] = 0$ . Hence the good quantum numbers for the eigenstates of  $\hat{L}^2$  are  $(l_1, l_2, l, m)$ , i.e., the corresponding quantum numbers of the operators  $\hat{L}_1^2, \hat{L}_2^2, \hat{L}^2$  and  $\hat{L}_z$ .

(b) Apply angular momentum addition theorem to obtain all eigenvalues of the following Hamiltonian

$$H = \hat{\mathbf{L}}_1 \cdot \hat{\mathbf{L}}_2 + \alpha \hat{L}_z$$

where  $\alpha$  is a constant,  $\hat{L}_z = \hat{L}_{z1} + \hat{L}_{z2}$  and the angular momentum quantum numbers of  $\hat{\mathbf{L}}_1$  and  $\hat{\mathbf{L}}_2$  are  $l_1 = 1$  and  $l_2 = 3/2$  respectively. Hint: use  $\hat{\mathbf{L}}^2 = \hat{\mathbf{L}}_1^2 + \hat{\mathbf{L}}_2^2 + 2\hat{\mathbf{L}}_1 \cdot \hat{\mathbf{L}}_2$ . 4. Two interacting spins (both with s = 1/2) have the following Hamiltonian

$$\hat{H} = \alpha \hat{\mathbf{S}}_1 \cdot \hat{\mathbf{S}}_2 \; ,$$

where  $\alpha$  is the coupling constant.

(a) Show that

$$\hat{\mathbf{S}}_1 \cdot \hat{\mathbf{S}}_2 = \frac{1}{2} \left( \hat{S}_1^+ \hat{S}_2^- + \hat{S}_1^- \hat{S}_2^+ \right) + \hat{S}_1^z \hat{S}_2^z \ .$$

(b) Use Angular Momentum Addition Theorem to determine its eigenvalues and their degeneracies.

(c) Determine the corresponding eigenstates in terms of the single-spin states  $|\uparrow\rangle$ and  $|\downarrow\rangle$ . Hint: Use results of Part (b) to construct four different two-spin states in terms of one-spin states  $|\uparrow\rangle$  and  $|\downarrow\rangle$  and use Part (a) to prove that they are indeed eigenstates of  $\hat{H}$ .

- 5. (a) State the complete Hund's rules.(b). Write down electron configuration of boron and derive its atomic terms. Which one corresponds to its ground state?(c) Repeat Part (b) for Carbon.
- 6. Consider a quantum system of two identical particles. In the independent-particle approximation, given n single-particle states  $[\phi_i(x), i = 1, 2, \dots, n; n \ge 2]$ , prove that there are n(n-1)/2 possible antisymmetric states and n(n+1)/2 symmetric states for the two-particle system. [Note: In proving this theorem, you may as well construct all these states, starting with n = 2, 3, etc., then extending to general n and proving it by induction.]